

VICINAL TRICARBONYL PRODUCTS FROM SINGLET OXYGEN REACTIONS.  
APPLICATION TO THE SYNTHESIS OF CARBACEPHAMS.

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**Abstract:** Vicinal tricarbonyl systems are readily formed by reacting  $\beta$ -dicarbonyl precursors with DMF acetal to form enamines which are then cleaved by photooxidation. This procedure may be applied to the formation of carbacephams.

We have recently described a procedure for the formation of  $\alpha$ -keto carbonyl derivatives by the conversion of ketones, lactones, esters and lactams<sup>1</sup> to their  $\alpha$ -enamino derivatives followed by oxidative cleavage with singlet oxygen. We have now shown that this method may be extended to the generation of vicinal tricarbonyl systems (**3**) from the corresponding  $\beta$ -dicarbonyl precursors (**1**) according to the general reaction in Scheme 1.<sup>2</sup> With the more reactive methylene groups in the  $\beta$ -diketonic substrates (**1**), it is possible to form the enamines (**2**) under very mild conditions, merely by mixing the  $\beta$ -dicarbonyl component with dimethylformamide dimethyl acetal at room temperature.<sup>3</sup>

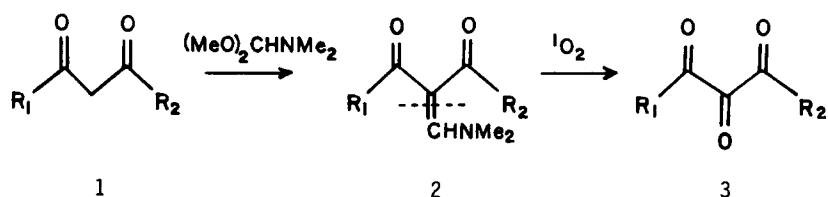
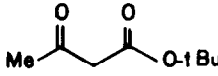
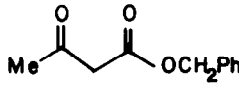
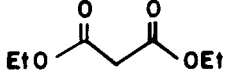
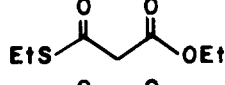
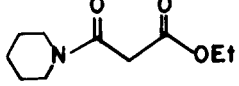


Table 1 provides information on the  $\beta$ -dicarbonyl substrates, the enamines and the products of photooxidation. Cleavage of **2** to **3** could also be accomplished by ozonolysis<sup>4</sup> but, as shown in the Table, the yields were generally poorer. The following typical experimental procedure is provided for entry #3 in the Table. Diethyl malonate (**1**,  $\text{R}_1=\text{R}_2=\text{OEt}$ ) (260 mg, 1.63 mmole) was stirred with dimethylformamide dimethyl acetal<sup>3</sup> (388 mg, 3.26 mmole) for 6h at room temperature. All volatile products were removed *in vacuo*. The residue was purified by rapid silica gel chromatography, yielding 334 mg (96%) of dimethylaminomethylene (**2**,  $\text{R}_1=\text{R}_2=\text{OEt}$ ). This product (313 mg, 1.45 mmole) was then

Table I

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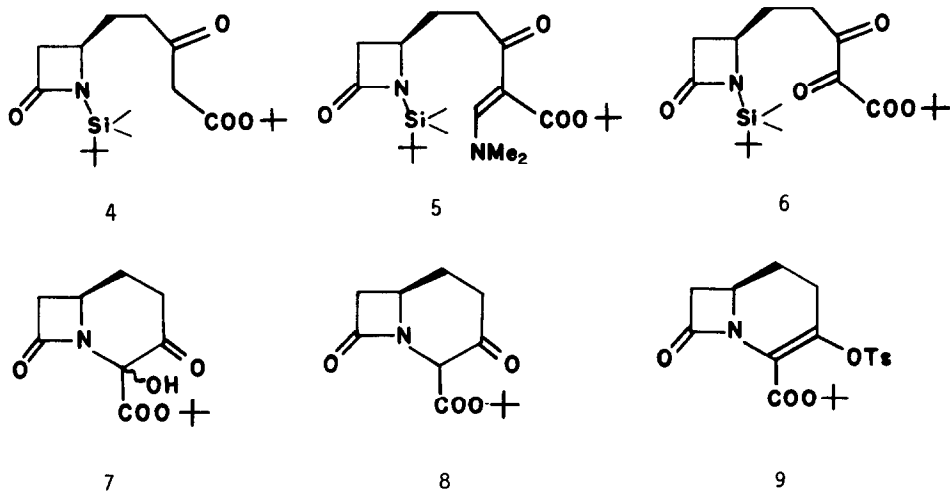
| Entry | $\beta$ -Dicarbonyl Precursor (1)   | Enamine (2) (%) | Tricarbonyl Product (3) (%) |       |
|-------|---|-----------------|-----------------------------|-------|
|       |   |                 | singlet oxygen              | ozone |
| 1     |  | 94              | 79                          | 55    |
| 2     |  | 95              | 83                          | 56    |
| 3     |  | 96              | 84                          | 94    |
| 4     |  | 75              | 65                          | 54    |
| 5     |  | 93 **           | 83                          | 76    |

\* Products were obtained in the hydrated form. They were identified by NMR, IR, and elemental analyses or high resolution mass spectroscopy.

\*\* In this case, reaction with DMF acetal took place at 54°C for 24 h.

dissolved in 20 ml of deuteriochloroform and photooxidized at 25°C for 20 h (BANT sensitizer).<sup>1</sup> The solution was then concentrated and the residue purified by silica gel chromatography to afford 235 mg (84%) of diethyl oxomalonate (3,  $R_1=R_2=OEt$ ) (in hydrated form).

This method for the *in situ* generation of vicinal tricarbonyl systems has useful application in synthesis as is illustrated in the formation of the carbacephem (8)<sup>5</sup>. The silylated chiral  $\beta$ -lactam derivative (4) formed by a route analogous to the Merck procedure<sup>6</sup>, was treated with N,N-dimethylformamide dimethyl acetal (25°C, 4h), forming the enamino ketone (5) (79%). Oxidation of (5) with singlet oxygen in  $DCCl_3$  as described earlier ( $^1O_2$ , BANT, 19°C, 4h)<sup>1</sup> gave the vicinal tricarbonyl derivative (6). The solvent, ( $DCCl_3$ ) was then replaced with  $CH_3CN$ , and desilylation achieved with HF-pyridine complex.<sup>7</sup> The acetonitrile solution was partitioned between ethyl acetate and  $H_2O$ , the ethyl acetate layer concentrated, and the oily residue taken up in dry  $CH_2Cl_2$ . Stirring the solution in the presence of activated molecular sieves (3Å) followed by silica gel chromatography yielded 7 as a separable mixture of two isomers (32% and 36%) differing in the



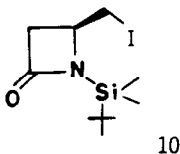
$\alpha$ -vs  $\beta$ -orientation of the newly formed hydroxyl group. Reduction of either of the isomers of **1** to a single product, (**2**), could be accomplished by treatment with trimethylsilyl iodide<sup>8</sup> (2.3 eq.,  $-40^{\circ}\text{C}$  to  $0^{\circ}\text{C}$ ,  $\text{CH}_2\text{Cl}_2$ ). Upon aqueous work-up ( $\text{K}_2\text{S}_2\text{O}_3$ ) and without further purification, **2** was treated with  $\text{Et}_3\text{N}$  and *p*-toluenesulfonyl anhydride to yield the enol tosylate (**3**) (45% overall yield from either the  $\alpha$ -OH or the  $\beta$ -OH form of **1**), identical with the product prepared by the Merck procedure<sup>9</sup>. Compound **3** has previously been converted to homothienamycin.<sup>9</sup>

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